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THE UNITED REPUBLIC OF TANZANIA NATIONAL EXAMINATIONS COUNCIL CERTIFICATE OF SECONDARY EDUCATION EXAMINATION

NOVEMBER 2015

032/2A

CHEMISTRY 2A (ACTUAL PRACTICAL A)

ADVANCE INSTRUCTIONS

1.0 IMPORTANT

- 1.1 GREAT CARE MUST BE TAKEN NOT TO DIVULGE THESE INSTRUCTIONS TO BOTH CANDIDATES AND UNAUTHORIZED PERSONS EITHER DIRECTLY OR INDIRECTLY.
- 1.2 NOTE THAT YOU WILL NOT BE ALLOWED TO OPEN THE ENVELOPE CONTAINING THE QUESTION PAPER BEFORE THE TIME OF COMMENCEMENT OF THE EXAMINATION.
- 1.3 24 HOURS ADVANCE INSTRUCTIONS FOR THIS PAPER WILL BE PROVIDED.
- In addition to the normal fittings and reagents in a normal CSE Chemistry laboratory, each candidate will require the following:

2.1 Apparatuses:

- 1 pipette (20 cm³ or 25 cm³)
- 1 burette (50 cm³)
- 3 titration flasks
- 2 beakers (250 or 300 cm³)
- 1 spatula
- 2 test tubes (pyrex)
- 2 Petri dishes/watch glasses

Measuring cylinder

- 1 Qualitative Analysis Guidance sheet.
- 100 cm³ beaker
- 1 stopwatch
- 1 heat source/burner
- 1 test tube holder
- 10 cm³ measuring cylinder

2.2 Chemicals:

- 4 g hydrochloric acid
- 3 g ammonium chloride
- 0.1 g potassium ferrocyanide
- 10 cm³ methyl orange indicator. 3 g lead (II) chloride (PbCl₂)
- 600 cm³ distilled water.
- 10 cm³ phenolphthalein (POP) indicator
- 3 g anhydrous sodium carbonate
- 0.1 g manganese dioxide (MnO₂)
- 3 g lead (II) nitrate (Pb(NO₃)₂). 2 g sodium thiosulphate (Na₂S₂O₃)

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